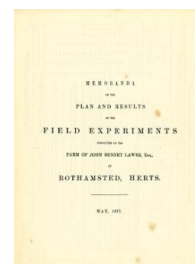


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ROTHAMSTED
RESEARCH

Memoranda of the Field Experiments at Rothamsted, May 1872



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Experiments on Oats; Geescroft Field

Rothamsted Research

Rothamsted Research (1873) *Experiments on Oats; Geescroft Field* ; Memoranda Of The Field Experiments At Rothamsted, May 1872, pp 5 - 5 - DOI: <https://doi.org/10.23637/ERADOC-1-236>

GEESCROFT FIELD.

EXPERIMENTS ON THE GROWTH OF OATS YEAR AFTER YEAR ON THE SAME LAND; WITHOUT MANURE, AND WITH DIFFERENT KINDS OF MANURE.
 Previous Cropping—1847 and 1848, Clover, Experimental Manures; 1849—1859, Beans, Experimental Manures; 1860, Fallow; 1861 and 1862, Wheat, Unmanured;
 1863, Fallow; 1864, Beans, Dunged; 1865, Wheat, Unmanured; 1866, Beans, Unmanured; 1867 and 1868, Wheat, Unmanured;
 First Experimental Oat Crop in 1869.

(Area under Experiment, $\frac{2}{3}$ acre).

PLOTS.	Manures, per Acre, per annum.	PRODUCE PER ACRE.											
		1ST SEASON, 1869.				2ND SEASON, 1870.				3RD SEASON, 1871.			
		Quantity.	Weight per Bushel.	Total Straw.	Dressed Corn.	Quantity.	Weight per Bushel.	Total Straw.	Dressed Corn.	Quantity.	Weight per Bushel.	Total Straw.	Dressed Corn.
1	Unmanured	Bushels. 36 $\frac{3}{4}$	lbs. 36 $\frac{3}{4}$	cwts. 19 $\frac{1}{2}$	Bushels. 16 $\frac{3}{4}$	lbs. 35	cwts. 9 $\frac{1}{2}$	Bushels. 20 $\frac{3}{4}$	lbs. 33 $\frac{1}{2}$	cwts. 11 $\frac{1}{2}$	Bushels. 20 $\frac{3}{4}$	lbs. 33 $\frac{1}{2}$	cwts. 11 $\frac{1}{2}$
2	{ 200 lbs. Sulphate Potass, 100 lbs. Sulphate Soda, 100 lbs. Sulphate Magnesia, and 3 $\frac{1}{2}$ cwt. Super-phosphate of Lime ⁽¹⁾	45	38 $\frac{1}{2}$	21 $\frac{1}{2}$	19 $\frac{1}{2}$	35 $\frac{1}{2}$	9 $\frac{1}{2}$	22	35 $\frac{1}{4}$	13 $\frac{1}{2}$	22	35 $\frac{1}{4}$	13 $\frac{1}{2}$
3	400 lbs. Ammonia-salts ⁽²⁾	56 $\frac{1}{2}$	37 $\frac{1}{2}$	36 $\frac{1}{2}$	30	34 $\frac{3}{4}$	17 $\frac{1}{2}$	57 $\frac{1}{2}$	36 $\frac{3}{4}$	40 $\frac{3}{4}$	57 $\frac{1}{2}$	36 $\frac{3}{4}$	40 $\frac{3}{4}$
4	{ 400 lbs. Ammonia-salts, 200 lbs. Sulphate Potass, 100 lbs. Sulphate Soda, 100 lbs. Sulphate Magnesia, and 3 $\frac{1}{2}$ cwt. Superphosphate	75 $\frac{1}{2}$	39 $\frac{1}{2}$	54	50 $\frac{3}{4}$	36	28 $\frac{1}{2}$	58 $\frac{3}{4}$	35 $\frac{3}{4}$	50	58 $\frac{3}{4}$	35 $\frac{3}{4}$	50
5	550 lbs. Nitrate of Soda ⁽³⁾	62 $\frac{1}{2}$	38 $\frac{1}{2}$	42 $\frac{3}{4}$	36 $\frac{1}{2}$	35 $\frac{1}{2}$	23	55	36 $\frac{3}{4}$	34 $\frac{3}{4}$	55	36 $\frac{3}{4}$	34 $\frac{3}{4}$
6	{ 550 lbs. Nitrate of Soda, 200 lbs. Sulphate Potass, 100 lbs. Sulphate Soda, 100 lbs. Sulphate Magnesia, and 3 $\frac{1}{2}$ cwt. Superphosphate	69 $\frac{3}{4}$	38 $\frac{1}{2}$	49 $\frac{1}{2}$	50	35 $\frac{3}{4}$	28 $\frac{3}{4}$	60 $\frac{1}{4}$	33 $\frac{3}{4}$	43 $\frac{3}{4}$	60 $\frac{1}{4}$	33 $\frac{3}{4}$	43 $\frac{3}{4}$

(1) "Superphosphate of Lime"—in all cases, made from 200 lbs. Bone-ash, 150 lbs. Sulphuric Acid sp. gr 1.7 (and water).
 (2) "Ammonia-salts"—in each case, equal part Sulphate and Nitrate of Ammonia of Commerce.
 (3) 550 lbs. Nitrate of Soda is reckoned to contain the same amount of Nitrogen as 400 lbs. "Ammonia-salts."